

1,3-CYCLOADDITIONS OF ALIPHATIC THIONE *S*-METHYLIDES TO DIMETHYL 2,3-DICYANO-
FUMARATE AND 2,3-DICYANOMALEATE; A TEST CASE FOR STERIC COURSE AND MECHANISM ¹

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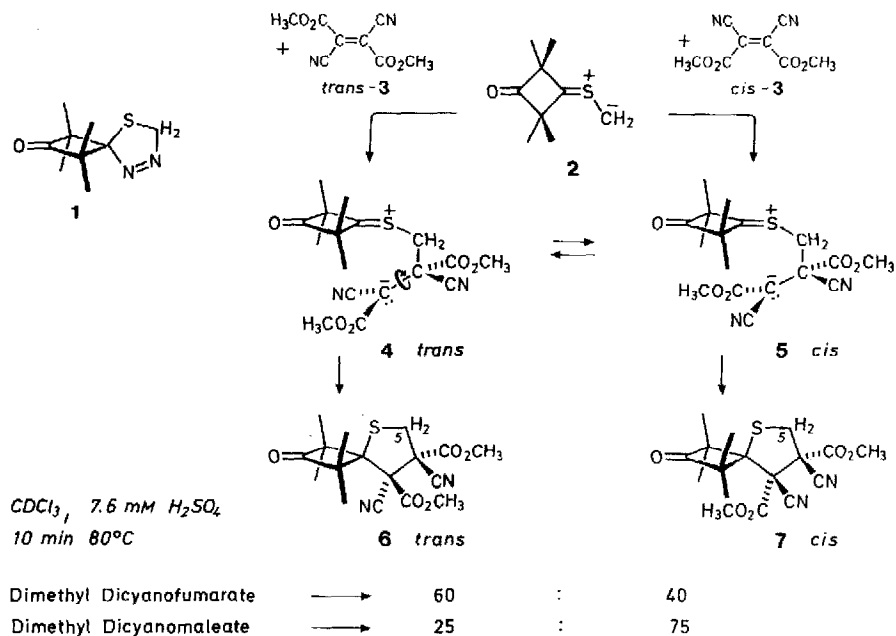
Summary The 1,3-cycloadditions of thione *S*-methylides 2 and 9 to dimethyl 2,3-dicyanofumarate and 2,3-dicyanomaleate are nonstereospecific. A preceding *cis,trans* isomerization of the unsaturated dipolarophiles catalyzed by the thiadiazolines (precursors of the thione *S*-methylides) had to be suppressed in order to clarify the stereochemical leakage *during* the cycloaddition.

The easily available spiro-1,3,4-thiadiazoline 1² extrudes N₂ at 40°C in a 1,3-dipolar cycloreversion (THF, $t_{1/2}$ 87 min); the thione *S*-methylide 2 generated is interceptible *in situ*. We reported on the nonstereospecific addition of 2 to the C=C bond of 2,3-dicyanofumaric ester (*trans*-3) affording *trans* and *cis*-adduct, 6 and 7, in the ratio of 52:48 (THF, 8 h 40°C, 90%);³ there was no evidence for the occurrence of *cis*-3. A great difference of the π -MO energies of the nucleophilic 1,3-dipole and the electrophilic dipolarophile as well as steric screening of *one* terminus of 2 were regarded as prerequisites for the unusual *two-step pathway* of 1,3-dipolar cycloaddition. The zwitterions 4 and 5 were proposed as attractive intermediates.^{3,4} In contrast, the combination of 2 both with fumaronitrile and maleonitrile are stereospecific with >99.9%; these cycloadditions are probably *concerted*.

In 1986 Gotoh, Padias, and Hall⁵ succeeded in preparing *dimethyl 2,3-dicyanomaleate* (*cis*-3) by isomerization of *trans*-3 photosensitized by 1 equiv of 1,4-dicyanobenzene. In our hands, the direct photoreaction in CH₂Cl₂ in an acid-washed duran reactor (high-pressure mercury arc) proved more convenient, sparing us the separation of the sensitizer. The irradiated solution contained *cis*-3/*trans*-3 = 91:9 and allowed the crystallization of 86% *cis*-3 with <2% *trans*-3. Several recrystallizations were required to diminish the share of the less soluble *trans*-3 to $\leq 0.2\%$.

The ¹H NMR analysis was based on the OCH₃ singlets of *trans*-3 (δ 4.03) and *cis*-3 (3.94) in CDCl₃; comparison with the ¹³C satellites of the major isomer in the high-field spectrum allowed quantitative determination of small percentages. Nucleophilic reagents like dispersed KF established an 88:12 *trans, cis* equilibrium of 3 (CDCl₃, 2 days).

It was observed that thiadiazoline 1 catalyzes a slow *cis,trans isomerization* of 3 at room temperature; not the thioether, but rather the *cis* azo group of 1 appears to be responsible. A configurational change of the dipola-



rophile 3 prior to the cycloaddition seriously interferes with the stereospecificity test. Indeed, after reacting 1 with 1.1 equiv of trans-3 or cis-3, the unconsumed 3 showed isomerization, little in the case of trans-3 and much for cis-3, due to the position of the equilibrium.

We noticed an inhibitory function of acids on the thiadiazoline catalysis. Optimal was 7.6 mM H_2SO_4 in CDCl_3 reproducibly obtained by shaking CDCl_3 ⁶ with 98% sulfuric acid and decanting after 2 days. 0.1 M Thiadiazoline 1 isomerized 0.12 M cis-3 in CDCl_3 within 250 min at 25°C to cis/trans = 81:19; a 98:2 ratio was observed in the presence of 7.6 mM H_2SO_4 . A second trick further assisted in curbing the undesired catalysis. Thiadiazoline 1 enters into competing pathways: *bimolecular* catalysis of the isomerization cis-3 \rightleftharpoons trans-3 and *unimolecular* extrusion of N_2 . Due to the negative ΔS^\ddagger of bimolecular reactions, the rate constant of the catalytic isomerization should rise less with increasing temperature than k_1 of the cycloreversion 1 \rightarrow 2 + N_2 .

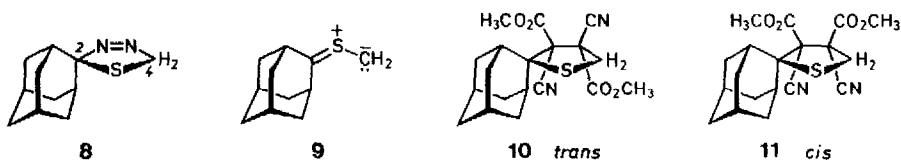
When 1 was reacted with 1.2 equiv of cis-3 in CDCl_3 , 7.6 mM in H_2SO_4 , for 10 min at 80°C, the 360 MHz ^1H NMR spectrum indicated less than 1% trans-3 in the unconsumed cis-3; in contrast, after the reaction at 40°C (10 h) the share of trans-3 was 10%. Due to fortunate analytical conditions, several isolated signals of thiolanes 6 and 7 were available for integration and comparison with weighed standards. The averaged ratios of trans and cis-thiolane observed in a series of experiments with trans-3 and cis-3 are shown in the formula scheme above. Thiolane yields were mostly 60-80%.

No isomerization, $\underline{6} \rightleftharpoons \underline{7}$, was noticed even under drastic conditions, 7 d in benzonitrile at 139°C. Reactants and products being configurationally stable, the substantial loss of stereochemical integrity leaves no doubt: rotation must take place in intermediates - the zwitterions $\underline{4}$ and $\underline{5}$ are likely candidates - before the thiolane rings are closed. Rotational equilibrium is not attained, and the stereochemical leakage was greater for the trans-zwitterion $\underline{4}$ (40%) than for the cis-form $\underline{5}$ (25%). A competition experiment revealed that $\underline{2}$ combined with trans- $\underline{3}$ 4.6 times faster than with cis- $\underline{3}$ (CDCl_3 , 80°C).

The suppression of the preceding cis,trans isomerization of $\underline{3}$ at 80°C in the presence of H_2SO_4 rules out an alternative pathway: rotation in the zwitterion and *dissociation* back to the reactants. For entropy reasons, more dissociation would be expected at 80°C than at 40°C. However, less isomerization of cis- $\underline{3}$ occurred at the higher temperature.

Diisopropyl 2,3-dicyanofumarate and 2,3-dicyanomaleate are less prone to the nucleophilic catalysis of the isomerization. In tests with 1 equiv of $\underline{1}$ in CDCl_3 at 25°C, no cis,trans isomerization of cis- $\underline{3}$ was detected after 270 min, with and without added H_2SO_4 . The advantages of lower sensitivity to nucleophiles and higher solubility are offset by analytical difficulties: the isopropyl signals of the cis,trans isomeric dipolarophiles overlap with those of the thiolanes, $\underline{6}$ and $\underline{7}$, $\text{OCH}(\text{CH}_3)_2$ instead of OCH_3 , i.e., dicyanofumarate cannot be discerned in the unconsumed dicyanomaleate.

Reactions of $\underline{1}$ with the diisopropyl esters were run in CDCl_3 , 7.6 M in H_2SO_4 , for 10 min at 80°C, and integration of the 5- H_2 signals of the thiolanes provided the same trans,cis adduct ratios as the dimethyl esters $\underline{3}$: 61:39 starting from diisopropyl dicyanofumarate and 25:75 from the cis isomer.



1,3,4-Thiadiazoline-2-spiro-2'-adamantane ($\underline{8}$) generates $\underline{9}$ at 45°C.⁷ According to our earlier report,³ $\underline{9}$ added to trans- $\underline{3}$ in THF furnishing trans and cis-thiolane, $\underline{10}$ and $\underline{11}$, in 59:41 ratio. Thiadiazoline $\underline{8}$ (0.1 M in CDCl_3) turned out to be a more active catalyst than $\underline{1}$ in the isomerization of cis- $\underline{3}$ (0.11 M): after 250 min at 25°C, cis- $\underline{3}$ /trans- $\underline{3}$ = 54:46 was measured; 73:27 in the presence of 7.6 mM H_2SO_4 testified to an insufficient protection.

In a series of experiments with 1.1 equiv of cis- $\underline{3}$ (7.6 mM H_2SO_4) in CDCl_3 , 10 min 80°C), the concentration of $\underline{8}$ was increased from 26 mM to 1540 mM in the expectation that a rising trans/cis thiolane ratio, $\underline{10}/\underline{11}$, would re-

flect the increase of isomerization of 3 prior to the cycloaddition. Ratios of 10/11 from 14:86 to 31:69 were monitored, but the amount of trans-3 in the un-consumed cis-3 did not regularly increase with the concentration of 8. The best experiment exhibited only 11% cis→trans isomerization in the remaining cis-3, and the thiolane ratio of 14:86 was the smallest. A ratio of 10/11 = 72:28 in experiments with trans-3 appears even more trustworthy, since the equilibrium trans-3 ⇌ cis-3 harbors only 12% cis isomer, and trans-3 exceeds cis-3 in dipolarophilic activity.

With shares of 28% cis-thiolane 11 from trans-3 and 14% (or less) trans-thiolane 10 from cis-3, the cycloadditions of *S*-methylide 9 present more retention of dipolarophile configuration than those of 2. The interaction of 8 with the less sensitive diisopropyl esters confirmed the magnitude: 23% cis-thiolane from dicyanofumarate and 15% trans-thiolane from dicyanomaleate.

The mechanistic conclusions of the earlier report ³ remain untouched. Differences between thiocarbonyl ylides 2 and 9 in their steric course of cycloaddition can be explained either by different rate ratios of rotation and cyclization of the zwitterionic intermediates or by varying participation of a concerted pathway (retention). The stereospecific (> 99.6%) cycloaddition of thiobenzophenone *S*-methylide to trans-3 ⁸ may be remembered. However, the non-stereospecificity observed for some other aliphatic thione *S*-methylides ⁴ calls for reexamination; the losses of dipolarophile configuration *before* and *during* the cycloaddition need to be distinguished.

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